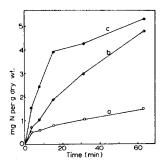
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Oscillatory transport of ammonium ions by Scenedesmus quadricauda

Chlorococcal algae of the genera Chlorella and Scenedesmus have been used for studying physiological processes in algae both under growth and nongrowth conditions. In search for a suitable medium for *Scenedesmus quadricauda* we have examined in some detail the uptake and release of various inorganic cations and anions by this species and their mutual interactions during transport^{1, 2}. In the present communication we wish to report some of the properties of the transport system for NH₄⁺ in this alga.

The experimental organism was Scenedesmus quadricauda (Turp) Bréb. strain Greifswald/15 of the collection of the Algological Laboratory at Třeboň. It was grown under static conditions³ at 26°, with fluorescent tube illumination (about 0.05 cal· cm⁻²·min⁻¹, Phar). The suspension was mixed and saturated with a mixture of air and CO2 (95:5, v/v) at a flow rate of 50 l/h per l of suspension. The nutrient solution used contained 40 mM urea, 5 mM MgSO₄·7 H₂O, 5 mM KH₂PO₄, microelements and EDTA. After growth the cells were centrifuged, washed and, at a density of some 3 % (referring to dry weight), were used for incubation experiments. NH₄HCO₃ at the appropriate concentration was added to the suspension at zero time, and 10-ml samples of the suspension were removed at 30-sec intervals, filtered through a Whatman No. I filter attached to a water pump and the filtrate analyzed. Where efflux was studied, the suspension was preincubated for 2 h with NH₄HCO₃ (300 mg N per l), cells were separated by filtration and resuspended in an NH₄+-free medium. The sampling procedure was the same as with influx. NH₄+ was determined in the filtrate colorimetrically by Nessler's method. The standard deviation of the estimation was 1.09%. It should be noted that the culture was not starved for nitrogen, the urea present in the medium being readily incorporated as such into the metabolic pathways of the cell4.

It was found that NH_4^+ is taken up from the medium by a temperature-dependent process (Fig. 1) against its concentration gradient (Fig. 2). It may be seen from Fig. 3 that the NH_4^+ taken up over a period of 2 h are lost again when cells



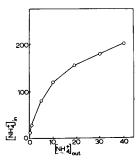


Fig. 1. Uptake of NH_4^+ by S. quadricauda at 5° (a), 15° (b) and 35° (c) under illumination from an initial concentration of 7 mM NH_4^+ .

Fig. 2. Relationship between external $\mathrm{NH_4^+}$ concentration and its intracellular steady-state level in S. quadricauda, determined from the decrease of $\mathrm{NH_4^+}$ concentration in the medium after 90 min of incubation and expressed in mmoles/l of total cell water.

are resuspended in an NH_4^+ -free medium and thus can be assumed to be present in a free form inside the cells. The maximum ratio of distribution attained at low concentrations of NH_4^+ was about 20:1, the half-saturation constant of uptake was 1.25 mM.

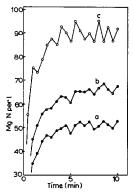


Fig. 3. Oscillatory efflux of $\mathrm{NH_4^+}$ from S. quadricauda, determined in the medium after preincubation of cells with $\mathrm{NH_4HCO_3}$ (300 mg N per ml) for 2 h and after transfer to an $\mathrm{NH_4^+}$ -free solution. a, at 5°, with 30 mg cells per ml; b, at 35°, with 30 mg cells per ml; c, at 35°, with 60 mg cells per ml.

The efflux of NH₄+ was subjected to graphical analysis and the efflux curve decomposed into at least three components. The fastest component of the efflux was found to be associated with an activation energy of 5.1 kcal/mole, the value being rather low and suggesting only minimum interaction with membrane components during the transport process. A peculiar observation was made when studying the efflux curve, which was found to display periodic changes in the concentration of NH₄+ present in the medium and, hence, in the cells. The oscillation amplitude was as much as 10 times greater than the standard deviation of the mean of NH4+ estimation and therefore considered to be significant. Moreover, the period of the oscillations was reproducible from one experiment to another, its mean value being 77 sec, with very little effect of temperatures between 5 and 35°. The amplitude, on the other hand, increased with temperature (Table I). It is noteworthy that application of 0.5 mM 2,4-dinitrophenol and of 5 µg/ml valinomycin had no effect on the oscillation frequency. Estimation of the pH of the external medium during NH₄+ outflow showed it to decrease from any preset value (between 4 and 8) by a slow, nonoscillatory process. Hence there does not appear to be a correlation between the movement of NH₄+ and H+.

TABLE I oscillation parameters of $\mathrm{NH_4}^+$ efflux at different temperatures

Тетр.	Period (sec)	Amplitude (µg N per ml cells)	Number of experiments
5° 15° 25° 35°	76	8.2	10
15°	79	10.7	10
25°	77	16.5	10
35°	80	23.0	30

In a series of preliminary experiments a practically identical type of oscillation was found in the efflux of K^+ from S. quadricauda. In their insensibility to temperature and inhibitors, the oscillations resemble those in the uptake of sugars and amino acids by the yeast Saccharomyces cerevisiae (prepared for publication) and suggest a more widespread mechanism to be involved. While it is rather unlikely that the NH_4^+ is pumped back and forth actively, the hypothesis presented is that rhythmic changes of cell volume or at least of a part of it are somehow implicated in the phenomenon.

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